

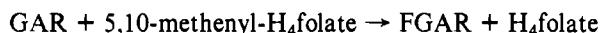
L(-)-10-Formyltetrahydrofolate Is the Cofactor for Glycinamide Ribonucleotide Transformylase from Chicken Liver[†]

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ABSTRACT: It is shown that L(-)-10-formyltetrahydrofolate serves as the cofactor for glycinamide ribonucleotide transformylase from chicken liver. The utilization of L(-)-10-formyl-H₄folate was not previously recognized, because L-(+)-10-formyl-H₄folate is an excellent competitive inhibitor

of the enzyme, $K_i = 0.75 \pm 0.07 \mu\text{M}$, and historically the transformylase assay was carried out with a mixture of diastereomers. The results are discussed in relation to the utilization of L-(+)-5,10-methenyl-H₄folate.

Buchanan and co-workers first characterized the GAR TFase¹ reaction in a crude preparation from chicken liver (Hartman & Buchanan, 1959). They showed that the cofactor is 5,10-methenyl-H₄folate¹ and that the overall reaction is



We have recently shown that a second enzyme, the trifunctional protein, is required in addition to GAR TFase to catalyze the reaction as written (Smith et al., 1980). The trifunctional protein apparently binds the 5,10-methenyl-H₄folate from solution and then "furnishes" a form of the cofactor to the GAR TFase binding site. We have also shown that a number of 10-formyl analogues (such as 10-formyl-5,8-deazafofolate) can bypass this requirement and bind directly to GAR TFase to formylate GAR in the presence or absence of the trifunctional protein (Smith et al., 1981). These results suggested to us that in the natural reaction the active form of the cofactor is not 5,10-methenyl-H₄folate but rather 10-formyl-H₄folate and led us to reinvestigate the direct utilization of 10-formyl-H₄folate as a cofactor.

The results of the present study show that the L(-)-10-formyl-H₄folate (the hydrolysis product of L-(+)-5,10-methenyl-H₄folate) is the true cofactor for GAR TFase. Severe inhibition by L-(+)-10-formyl-H₄folate in chemical preparations of L(±)-10-formyl-H₄folate had previously masked this activity. These results are in agreement with the report that *Escherichia coli* GAR TFase also utilizes 10-formyl-H₄folate as its cofactor (Dev & Harvey, 1978)

Experimental Procedures

Materials. GAR TFase, the trifunctional protein, L(±)-5,10-methenyl-H₄folate, and (α,β)GAR were prepared as described previously (Smith et al., 1980). H₂folate was prepared according to the procedure of Blakley (1960), and L-(-)-H₄folate was produced from the H₂folate with H₂folate reductase (Mathews & Huennkens, 1960) and purified on DEAE-cellulose eluting with a 0.2–0.7 M NH₄HCO₃, pH 8.0, linear gradient (Curthoys et al., 1972). L-(+)-5,10-

Methenyl-H₄folate was synthesized from L(-)-H₄folate according to the method of Rowe (1968). L(-)-5,10-Methenyl-H₄folate was a generous gift of the Southern Research Institute. Hydrolysis of (+)- and (-)-5,10-methenyl-H₄folate gave (-)- and (+)-10-formyl-H₄folate, respectively.

Methods. All solutions, except enzymes which were always added in small amounts, were thoroughly degassed by alternately evacuating and saturating with argon with at least eight exchanges. All reactions were performed in argon-flushed, serum cap sealed tubes or cuvettes. Additions and transfers were made through syringes.

Buffer-catalyzed hydrolysis of 5,10-methenyl-H₄folate to 10-formyl-H₄folate was performed at room temperature in 40 mM potassium phosphate and 0.1 mM EDTA, pH 6.8. The reaction was allowed to proceed in the dark for 2 h, at which time it was judged complete by the product UV spectrum (Rabinowitz, 1963). The solution was used for that day's work, after which recyclization to 5,10-methenyl-H₄folate in 0.1 M HCl indicated very little decomposition.

GAR TFase assays were performed at pH 6.8 in 50 mM maleate buffer containing 0.1 mM EDTA, 0.259 mM (α,β)-GAR, and varying amounts of 5,10-methenyl- or 10-formyl-H₄folate. The Bratton and Marshall assay for diazotizable amine was performed as previously described (Smith et al., 1980). Spectrophotometric assays were performed in a 1-mL reaction volume in a serum cap sealed cuvette. Absorbance changes were monitored at either 298 or 312 nm in a Gilford 240 spectrophotometer equipped with constant temperature and automatic recording accessories. Assay solutions were preincubated for 10 min at 35 °C in the absence of cofactor, and the reaction was initiated by adding the latter in a small volume. No difference was observed in the 10-formyl-H₄folate activity with GAR TFase whether enzyme or cofactor was used to start the reaction.

The Cleland programs for linear competitive and noncompetitive inhibition were used to analyze inhibition data (Cleland, 1967).

Results

The Bratton–Marshall diazotizable amine assay, typically used with GAR TFase, is a point assay requiring an acid quench and a four-step color development. For determination of the entire reaction time course, it is cumbersome and time consuming; thus, a spectrophotometric assay was sought.

The spectral change for the buffer-catalyzed hydrolysis of 5,10-methenyl-H₄folate to 10-formyl-H₄folate in 40 mM potassium phosphate, pH 6.8, is shown in Figure 1. The reaction shows an isosbestic point at 312 nm. At this wavelength, the difference in molar absorptivity between H₄folate and either

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¹ Abbreviations used: AICAR, 5-aminoimidazole-4-carboxamide ribonucleotide; GAR, glycinamide ribonucleotide; H₄folate, tetrahydrofolate; TFase, transformylase; FGAR, formylglycinamide ribonucleotide; trifunctional protein, 5,10-methenyl-, 5,10-methylene-, 10-formyl-H₄folate synthetase (combined). The (+), (-), or (±) notation is the conventional optical rotation designation for a given cofactor species.

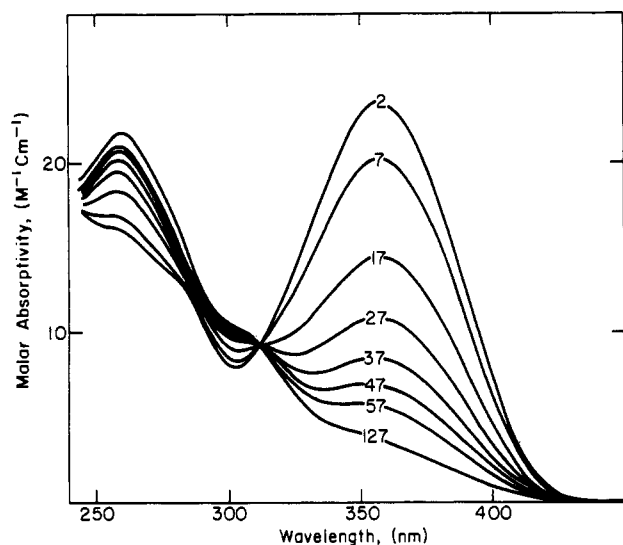


FIGURE 1: Buffer-catalyzed hydrolysis of 5,10-methenyl-H₄folate. The reaction was performed at 25 °C in 40 mM potassium phosphate buffer, pH 6.8. Times recorded on the spectra are in minutes.

Table I: Utilization of L(±)-10-Formyl- and L(±)-5,10-Methenyl-H₄folate by GAR TFase

| cofactor | [cofactor] (μM) | $v_0 \Delta A_{312}/\text{min}^a$ | |
|--|--------------------|-------------------------------------|------------------------------------|
| | | without trifunctional protein | with tri- functional protein |
| L(±)-10-formyl-H ₄ folate | 50 | 0.0030 ^b | 0.0030 |
| L(±)-10-formyl-H ₄ folate | 25 | 0.0027 ^b | 0.0029 |
| L(±)-5,10-methenyl-H ₄ folate | 50 | 0.0045 ^c | 0.019 |
| L(±)-5,10-methenyl-H ₄ folate | 25 | 0.0032 ^c | 0.0165 |

^a The enzymes were preincubated in the absence of cofactor for 10 min. For the GAR TFase enzyme, 2.5 μL of a solution of a 0.712 mg/mL GAR TFase and 0.054 mg/mL trifunctional protein was used. For the trifunctional protein, 5.0 μL of 0.2 mg/mL solution was used. ^b Adding trifunctional protein to the running assay had no effect. ^c Adding trifunctional protein to the running assay gave the expected activation within the mixing time.

5,10-methenyl-H₄folate or 10-formyl-H₄folate is 12 000 M⁻¹ cm⁻¹. Thus, the formylation of GAR with 5,10-methenyl-H₄folate can be followed with no interference from the 5,10-methenyl-H₄folate hydrolysis reaction. Similarly, the formylation from 10-formyl-H₄folate alone can be followed at 298 nm, as shown for AICAR TFase by Black et al. (1978): $\Delta\epsilon_{298} = 19\,700 \text{ M}^{-1} \text{ cm}^{-1}$. Using these spectral changes and the Bratton-Marshall technique, we have reinvestigated the cofactor requirement of GAR TFase from chicken liver.

It has previously been assumed that any GAR TFase activity found with 10-formyl-H₄folate as the cofactor was due to residual 5,10-methenyl-H₄folate in solution (Caperelli et al., 1980). However, we have shown that the GAR TFase activity with 5,10-methenyl-H₄folate requires the presence of the trifunctional protein while the activity with 10-formylfolate analogues does not (Smith et al., 1980, 1981); thus, if 10-formyl-H₄folate utilization is the same in the presence or absence of the trifunctional protein, the activity cannot result from residual 5,10-methenyl-H₄folate. The results of four assays with L(±)-10-formyl-H₄folate compared with similar assays with L(±)-5,10-methenyl-H₄folate are shown in Table I. It can be seen that there is identical utilization of the 10-formyl-H₄folate with or without the trifunctional protein

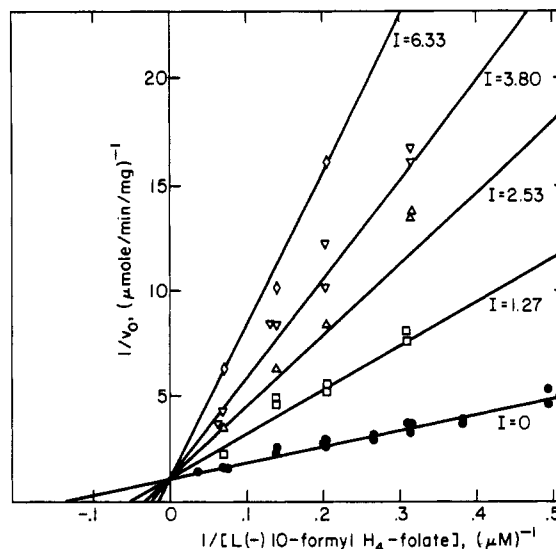


FIGURE 2: Double-reciprocal plot of the utilization of L(-)-10-formyl-H₄folate by GAR TFase and its inhibition by L(+)-10-formyl-H₄folate. Assay performed at 35 °C in 50 mM maleate, 0.1 mM EDTA, 0.259 mM (α,β)GAR, and 2.0 μg of complex. ΔA_{312} was recorded. Inhibitor concentrations are μM.

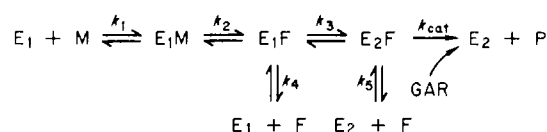
whereas efficient use of the 5,10-methenyl-H₄folate as a cofactor shows a requirement for the trifunctional protein. The V_{\max} for L(±)-10-formyl-H₄folate relative to L(±)-5,10-methenyl-H₄folate is 14% (0.080 compared to 0.59 μmol min⁻¹ mg⁻¹ for a typical enzyme complex preparation). It could be argued that the failure of the trifunctional protein to increase the utilization of the L(±)-10-formyl-H₄folate by GAR TFase is due to slow activation. However, no increase over the initial velocity with the 10-formyl species is observed during the assay period (up to 15 min) whereas activation of the 5,10-methenyl-H₄folate utilization occurs within the 15–20 s required for adding and mixing the last reagent (cofactor or trifunctional protein).

The assay was then repeated with L(-)-10-formyl-H₄folate, and it was found that the rate was identical with the rate with L(+)-5,10-methenyl-H₄folate (V_{\max} L(+)-methenyl-H₄folate = $0.91 \pm 0.05 \mu\text{mol min}^{-1} \text{ mg}^{-1}$ and L(-)-formyl-H₄folate = $0.95 \pm 0.07 \mu\text{mol min}^{-1} \text{ mg}^{-1}$ for the same complex used above). Again the trifunctional protein had no effect on the rate with 10-formyl-H₄folate but was required for L(+)-5,10-methenyl-H₄folate utilization, activating within the assay dead time (15–20 s). The reason for the large difference between the utilization of L(±)-10-formyl-H₄folate and L(-)-10-formyl-H₄folate is shown in Figure 2 where the competitive inhibition due to L(+)-10-formyl-H₄folate is displayed. It can be seen that the wrong diastereomer of 10-formyl-H₄folate severely inhibits GAR TFase, and from this plot, a K_i of $0.75 \pm 0.07 \mu\text{M}$ for L(+)-10-formyl-H₄folate was calculated, compared to a K_m of $6.76 \pm 0.65 \mu\text{M}$ for L(-)-10-formyl-H₄folate.

An attempt was made to fit the data in Figure 2 to a non-competitive scheme (Cleland, 1967). However, the K_{II} (intercept derived or noncompetitive K_I) calculated is negative, $-3.74 \pm 1.5 \mu\text{M}$. Thus, a noncompetitive model does not fit, and the competitive nature of the inhibition by L(+)-10-formyl-H₄folate is substantiated.

The K_m for L(-)-10-formyl-H₄folate in the absence of the trifunctional protein is $5.94 \pm 0.77 \mu\text{M}$. Thus, the trifunctional protein affects neither the K_m nor the V_{\max} of L(-)-10-formyl-H₄folate. The K_m for L(+)-5,10-methenyl-H₄folate was recalculated by employing this assay and gave $K_m = 5.76 \pm 0.60 \mu\text{M}$.

Scheme I



Discussion

The results show that L(-)-10-formyl-H₄folate is the active cofactor for GAR TFase in the absence of the trifunctional protein. This was not observed previously, because L(+)-10-formyl-H₄folate is surprisingly an excellent competitive inhibitor ($K_i = 0.75 \mu\text{M}$); thus, the presence of this diastereomer of the cofactor in L(±)-10-formyl-H₄folate formed by chemical reduction caused the reduced rate found with L(±)-10-formyl-H₄folate compared to L(±)-5,10-methenyl-H₄folate. (The latter is selectively hydrolyzed to the L(-)-10-formyl-H₄folate by the cyclohydrolase activity present in earlier preparations of GAR TFase.) Consequently the slow rate of (±) cofactor utilization in a solution at pH 6.8 where the cofactor concentrations are equilibrated (10-formyl-H₄folate:5,10-methenyl-H₄folate in the ratio 1:0.15; Poe & Benkovic, 1980) was presumed to originate from the residual 5,10-methenyl-H₄folate in the solution.

The activity of GAR TFase in the presence of the trifunctional protein when assayed with L(±)-5,10-methenyl-H₄folate is lower than that with L(+)-5,10-methenyl-H₄folate (0.59 compared to 0.91 $\mu\text{mol min}^{-1} \text{mg}^{-1}$). However, it cannot be determined whether this is due to slight inhibition from the L(-)-5,10-methenyl-H₄folate or from the small amount of L(+)-10-formyl-H₄folate which is certain to arise from buffer-catalyzed hydrolysis.

The probable role of the trifunctional protein in the GAR TFase-trifunctional protein mixture is outlined in Scheme I, where $E_1 \equiv$ cyclohydrolase activity of the trifunctional protein, $E_2 \equiv$ GAR TFase, $M \equiv$ 5,10-methenyl-H₄folate, $F \equiv$ 10-formyl-H₄folate, $P \equiv$ FGAR, and k_i 's are net rate constants. The evidence for postulating this scheme is that 10-formyl-H₄folate is used by GAR TFase in the absence of the trifunctional protein whereas 5,10-methenyl-H₄folate use requires active trifunctional protein.

In one limit, solutions M and F are at chemical equilibrium, and 10-formyl-H₄folate from solution is the true cofactor. This is met by the condition k_i ($i = 1, 2$, and 4) $\gg k_3$. In the present preparation, the cyclohydrolase has a specific activity of 2.80 units/mg at 15 μM L(+)-5,10-methenyl-H₄folate. Thus, the 2 μg of complex used in the assays would produce 1.9 μM 10-formyl-H₄folate in the 20-s mixing time of the assay. Consequently the GAR TFase velocity should have a pronounced lag while the 10-formyl-H₄folate concentration builds toward 13 μM . Further, the apparent K_m calculated from the initial velocity data with 5,10-methenyl-H₄folate should be greater than that found for 10-formyl-H₄folate since the initial

10-formyl-H₄folate concentration is less than the 15 μM L(+)-5,10-methenyl-H₄folate added. Neither of these criteria is fulfilled since the initial velocity at all 5,10-methenyl-H₄folate concentrations is constant and $K_m[\text{L}(+)-5,10\text{-methenyl-H}_4\text{folate}] = K_m[\text{L}(-)-10\text{-formyl-H}_4\text{folate}]$. Thus, the role of the trifunctional protein is not simply to generate active cofactor. Additional evidence inferring direct channeling between the two proteins was furnished by the demonstration of GAR TFase and trifunctional protein cross-linking, of their association in the presence of GAR, and of the increased rate of FGAR synthesis commencing with formate compared to that with 5,10-methenyl-H₄folate (Smith et al., 1980).

The relative values of k_i ($i = 1-5$) will determine the ratio of cofactor channeling to that dissociating to solution and influence K_m . However, the similarity in K_m values for L(+)-5,10-methenyl-H₄folate and L(-)-10-formyl-H₄folate does not readily follow from Scheme I without compensation within the values of k_i since two differing routes are utilized to saturate E_2F .

With the results presented here for GAR TFase and those of Hartman & Buchanan (1959) for AICAR TFase, it can be seen that both of the transformylases involved in purine biosynthesis employ L(-)-10-formyl-H₄folate as the common cofactor. The advantage of a direct formyl transfer agent rather than an amidinium donor in formylation may reside partially in the simplification afforded by controlling at the active site the bidirectional partitioning of a putative tetrahedral intermediate derived from the 10-formyl cofactor rather than the tridirectional routes open to a transient orthoamide species generated from the 5,10-methenyl cofactor.

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